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Original Article

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Bio- Mathematical Variations in Paralysis Using Flame Atomic Absorption Spectroscopy

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Abstract

We have tried to see the impact of trace elements in paralytic patients. A detailed statistical analysis was applied to the data in the present work. It is found that the data available with the flame atomic absorption spectroscopy method are higher in paralytic patients in comparison to the normal healthy controls. We have also evaluated the correlations between two elements and regression equations with regression coefficients also. Different pattern was seen in all the trace elements.

Keywords: Regression equations; Regression coefficients and paralytic patients.

1. Introduction

Trace elements belong, like the vitamins, the essential amino acids and essential fatty acids to the elements human beings. These are in need to function properly and the human system is unable to produce from food. These elements are crucial parts of many kinds of biochemical conversions in the body like co-enzyme reactions. Excess and deficiency of elements can lead to a malfunctioning of several organsystems of human body as a whole. Some other elements and toxic metals can influence the need for a traceelement. Consequently ratios are sometimes also important.

Minerals are important to our health. These are inorganic chemicals and are not attached with carbon atom. Minerals and trace minerals can be differentiated easily. These trace minerals are called trace elements. If the cellular body requires less than 100 mg of eleven minerals is labeled a trace element and if accepts more than this level is labeled a mineral.

Trace elements are required in quantities of few milligrams or micrograms per day. A study of relationships of minerals with human health is very necessary and important. A balance of the level for minerals in every organ, tissue and cell of the human body may lead to a good health.

Minerals comprise only a fraction of total body weight. These are crucial for many functions of the human body. These include transporting oxygen, normalizing the central nervous system (CNS) and simulating growth, maintenance and repair of tissues and bones [1].

Human body contains lot of element out of 25 are divided into mainly three categories such as major, minor components and trace elements. A brief description but elements is given here. Zumkley [2], stated that clinical

experimental and epidemiological studies indicate that a large number of trace elements may be involved in the etiology of different human body disorders. Both increased levels of any trace element or reverse of this can influence the development of disease. On an average, a young person requires one hundred to one hundred fifty milligram of copper daily. Copper in higher concentration is found in liver, hair, muscle and lung [3]. The binding of zinc to amino acids and serum protein was studied by Prasad and Oberleas [4]. McCance and Widdowson [5], have concluded from their research and have reported that once the iron was absorbed by the human body, its excretion was very minimal and was not controlled either by gastrointestinal tract or the kidneys. McCance and Widdowson [5], have studied the magnesium content in human body and found a range of magnesium, which stated from 22.7 to 35.0 meq/ kg weight of tissues. Magnesium reaches us in many forms. Many authors [6-8] have reported that the ingested calcium mixes with digestive juice calcium in the proximal small intestine from where it is absorbed by a mechanism Human body contains electrolytes. The balance of the electrolytes is essential for function of cells and organs of human beings. Main electrolytes which are measured in the blood by the doctors are sodium, potassium, chloride and bicarbonate.

2. Materials and Methods

Blood sample of paralytic patients along with normal healthy control were collected from the Department of Neurology, Safdarjang Hospital, New Delhi-110016 after the approval of ethical committee of the hospital. 10 ml freshly drawn blood from each patient was collected in clean and dry test tube without any anti-coagulant. The test tube was kept for 45 minutes at room temperature $(22 \pm 2^{\circ}C)$ for the formation of clot. Sera of different patients were separated by centrifugation at 1500 r.p.m. upto15 minutes and were collected in screw capped test tubes.

The atomic absorption spectral estimation of the serum samples from normal persons and paralytic patients were carried out on atomic absorption spectrophotometer Model No. AA- 6300 of Shimadzu Japan, at Deptt. of Environmental study University of Delhi 110007.

3. Mathematical and Statistical Formulation

We would like to add here that simple formulae of statistics and mathematics have been used in the present work. We used different types of mathematical and statistical software for multiple and partial correlation coefficients in the study. We have calculated all multiple correlations with the help of determinant theory only [9].

Yule's Notation: If we consider a distribution involving 'n' random variables $X_{1, X_2, X_3, \dots, X_n}$. Then the equation of the plane of regression of X_1 on X_2, X_3, \dots, X_n is given by

The constants b' in Equation (1) are determined by the principle of least squares, i.e. ,by minimizing the sum of the squares of the residuals ,viz.,

The sum of the squares of residuals is given by

The summation being extended to the given values (N in number) of the variables. Here we make N observations on each of the variables X_1 on X_2, X_3, \dots, X_n .

The normal equations for estimating $b_{12,34\dots n}$ and $b_{13,24\dots n}$

Using the principle of least squares, the normal equations for estimating the (n-1),b's are :

$$\frac{\partial S}{\partial b_{12,34\dots n}} = 0 = -2\sum X_2 \left(X_1 - b_{12,34\dots n} X_2 + b_{13,24\dots n} X_3 + \dots + b_{1n,23\dots (n-1)} X_n \right) \dots \dots \dots [3]$$

$$\frac{\partial S}{\partial b_{13,24\dots n}} = 0 = -2\sum X_3 \left(X_1 - b_{12,34\dots n} X_2 + b_{13,24\dots n} X_3 + \dots + b_{1n,23\dots (n-1)} X_n \right) \dots \dots \dots [4]$$

$$\frac{\partial S}{\partial b_{1n,23\dots(n-1)}} = 0 = -2\sum X_n \left(X_1 - b_{12,34\dots n} X_2 + b_{13,24\dots n} X_3 + \dots + b_{1n,23\dots(n-1)} X_n \right) \dots [5]$$

which on simplification give

$$r_{12}\sigma_{1}\sigma_{2} = b_{12.34...n}\sigma_{2}^{2} + b_{13.24...n}r_{23}\sigma_{2}\sigma_{3} + \dots + b_{1n.23...(n-1)}r_{2n}\sigma_{2}\sigma_{n} \dots [7]$$

$$r_{13}\sigma_{1}\sigma_{3} = b_{12.34...n}r_{23}\sigma_{2}\sigma_{3} + b_{13.24...n}\sigma_{3}^{2} + \dots + b_{1n.23...(n-1)}r_{3n}\sigma_{3}\sigma_{n} \dots [8]$$

$$r_{1n}\sigma_{1}\sigma_{n} = b_{12.34...n}r_{2n}\sigma_{2}\sigma_{n} + b_{13.24...n}r_{3n}\sigma_{3}\sigma_{n} + \dots + b_{1n.23...(n-1)}\sigma_{n}^{2} \dots [9]$$
Hence the eliminant of b's between Eqn.[1], Eqn.[[7], Eqn.[[8] and Eqn.[[9] is

Dividing $C_{1,}C_{1,}C_{1,}\dots,C_{n}$ by $\sigma_{1},\sigma_{2},\dots,\sigma_{n}$ respectively and R_{1},R_{1},\dots,R_{n} by $\sigma_{2},\sigma_{3},\dots,\sigma_{n}$ respectively, we get.

and ω_{ij} is the cofactor of the element in the ith row and jth column of ω , we get from Eqn. [11]

Equation [13] is the required equation of the plane of regression of X_1 on X_2 , X_3 , X_n Equation [11] can be written as :

$$X_{1} = -\frac{\sigma_{1}}{\sigma_{2}} \cdot \frac{\omega_{12}}{\omega_{11}} X_{2} - \frac{\sigma_{1}}{\sigma_{3}} \cdot \frac{\omega_{13}}{\omega_{11}} X_{3} - \dots - \frac{\sigma_{1}}{\sigma_{n}} \cdot \frac{\omega_{1n}}{\omega_{11}} \dots$$
[14]

ComparingEqn. [13] with Eqn. [1], we get

$$b_{1n.23\dots(n-1)} = -\frac{\sigma_1}{\sigma_n} \cdot \frac{\omega_{l_n}}{\omega_{l_n}}$$

4. Results

Table-1. Trace elemental determinations from plasma of paralytic patients and normal controls								
S.N.	FACTOR	Zn in ppm	Cu in ppm	Fe in ppm	Na in ppm	K in ppm	Ca in ppm	Mg in ppm
1	Р	4.379	0.227	4.775	143.00	3.79	2.36	0.26
2	Р	2.346	0.019	0.225	145.00	5.19	3.03	0.35
3	Р	1.392	0.149	1.093	132.00	5.39	3.95	3.92
4	Р	2.850	0.164	0.164	141.00	5.38	2.17	1.24
5	Р	0.534	0.167	6.184	142.00	3.89	3.17	2.51
6	Р	3.871	0.202	0.282	144.00	3.69	4.38	0.31
7	Р	0.531	0.208	2.598	128.00	4.19	4.62	3.10
8	Р	0.902	0.112	0.548	138.00	3.99	4.86	2.79
9	Р	0.524	0.093	0.117	139.00	4.29	4.40	2.49
10	Р	0.776	0.104	2.776	140.00	3.99	4.41	1.49
11	Р	0.547	0.162	0.295	143.00	4.19	4.62	3.49
12	Р	0.332	0.122	0.237	140.00	3.89	3.96	3.79
13	Р	0.894	0.216	0.956	144.00	4.29	2.39	1.99
14	Р	1.455	0.193	0.789	143.00	4.19	3.99	1.35
15	Р	4.379	0.113	2.132	139.0	4.79	3.62	2.19
16	N	1.493	0.125	0.702	135.00	3.59	2.47	0.49
17	N	0.801	0.073	1.142	131.00	3.79	2.46	0.59
18	N	2.134	0.125	1.145	137.00	3.71	2.97	0.60
19	N	0.571	0.112	0.451	136.00	3.61	2.79	0.75
20	N	0.584	0.117	0.162	141.00	3.91	1.92	0.79
21	N	0.847	0.115	0.058	142.00	3.71	4.59	0.54
22	N	1.286	0.156	0.612	137.00	3.81	4.56	0.62
23	N	0.023	0.158	5.788	135.00	3.61	4.53	0.73
24	N	1.491	0.208	0.586	137.00	3.62	4.29	0.84

Table-1. Trace elemental determinations from plasma of paralytic patients and normal controls

Table-2. Mean and standard deviations in normal and paralytic patients

ELEMENT	(Mean±s.d) For paralysis	(Mean±s.d) For Normal	
Zn	1.6076 ± 1.272	1.025 ± 0.598	
Cu	0.150 ± 0.054	0.132 ± 0.035	
Fe	1.54 ± 1.78	1.182 ± 1.665	
Na	140.06 ± 4.49	136.77 ± 3.08	
K	4.34 ± 0.551	3.706 ± 0.104	
Ca	3.72 ±0.867	3.39 ± 1.017	
Mg	2.08 ± 1.18	0.6611 ± 0.113	

Table-3. Regression c equations of trace elements in normal controls

 $Zn = -0.122749456 \quad Cu + \quad 3.455056498 \quad Mg - \quad 1.111363848 \cdot 10^{-2} \quad Ca - \quad 2.20821668 \cdot 10^{-3} \text{ Fe} + \quad 2.099284772 \cdot 10^{-1} \text{ Na} + 6.147789476 \cdot 10^{-2} \quad K + \quad 7.64806502 \cdot 10^$

 $\label{eq:cu} \begin{array}{c} Cu = 16.68819082 \quad Mg - \ 0.171807372 \quad Ca - \ 3.66940813 \cdot 10^{-2} \quad Fe + \ 5.342786305 \cdot 10^{-1} \quad Na - \ 2.528407205 \cdot 10^{-1} \quad K- \ 3.817045778 \quad Zn + \ 5.44521922 \end{array}$

 $Mg = 3.832067636 \cdot 10^{-3} \text{ Cu} + 1.321879048 \cdot 10^{-3} \text{ Fe-6.197427003} \cdot 10^{-2} \text{ Na} + 1.729654479 \cdot 10^{-2} \text{ K} + 1.946154288 \cdot 10^{-1} \text{ Zn} + 3.022898595 \cdot 10^{-2} \text{ Cu} - 4.194125922 \cdot 10^{-2}$

Ca= -0.223195912 Fe-9.523792534·10⁻¹ Na+ 1.209076357·10⁻¹ K- 3.62305649 Zn- 1.801162277 Cu+ 22.17838366Mg+ 36.14288887

Fe= 8.436242872 Na+ 0.604444228 K- 3.476492121 Zn- 1.857752224 Cu+ 36.94614822 Mg- 1.077871426 Ca+ 104.0512515

 $Na= 1.320701459 \cdot 10^{-2} \text{ K} + 4.109606214 \cdot 10^{-1} \text{ Zn} + 3.363480124 \cdot 10^{-2} \text{ Cu} - 2.153863417 \text{ Mg} - 5.718997936 \cdot 10^{-3} \text{ Ca} + 1.049006948 \cdot 10^{-2} \text{ Fe} + 2.212113012$

Table-4. Regression equations of trace elements in paralysis

Table-4. Regression equations of trace elements in pararysis
$Zn = -6.134659462 \cdot 10^{-1} Cu + 3.757697369 Mg - 2.261481115 \cdot 10^{-2} Ca - 7.450389233 \cdot 10^{-2} Fe + 3.716497609 \cdot 10^{-1} Na + 3.757697369 Mg - 2.261481115 \cdot 10^{-2} Ca - 7.450389233 \cdot 10^{-2} Fe + 3.716497609 \cdot 10^{-1} Na + 3.757697369 Mg - 2.261481115 \cdot 10^{-2} Ca - 7.450389233 \cdot 10^{-2} Fe + 3.716497609 \cdot 10^{-1} Na + 3.757697369 Mg - 2.261481115 \cdot 10^{-2} Ca - 7.450389233 \cdot 10^{-2} Fe + 3.716497609 \cdot 10^{-1} Na + 3.757697369 Mg - 2.261481115 \cdot 10^{-2} Ca - 7.450389233 \cdot 10^{-2} Fe + 3.716497609 \cdot 10^{-1} Na + 3.757697369 Mg - 2.261481115 \cdot 10^{-2} Ca - 7.450389233 \cdot 10^{-2} Fe + 3.716497609 \cdot 10^{-1} Na + 3.757697369 Mg - 2.261481115 \cdot 10^{-2} Ca - 7.450389233 \cdot 10^{-2} Fe + 3.716497609 \cdot 10^{-1} Na + 3.757697369 Mg - 3.75769769 Mg - 3.757697769 Mg - 3.757697769 Mg - 3.757697769 Mg - 3.7576977697769 Mg - 3.757697769 Mg - 3.7576977697769 Mg - 3.7576977697769 Mg - 3.7576977697769 Mg - 3.75769776977697769 Mg - 3.7576977697769776977697769776977697769777697769777697769777777$
$1.681523712 \cdot 10^{-1} \text{ K} + 10.73647994$
$Cu = 6.573949664 \text{ Mg} + 9.060178901 \cdot 10^{-3} \text{ Ca-} 1.279464174 \cdot 10^{-2} \text{ Fe} + 6.442408358 \cdot 10^{-1} \text{ Na} + 5.982430704 \cdot 10^{-2} \text{ K} - 1.279464174 \cdot 10^{-2} \text{ Fe} + 6.442408358 \cdot 10^{-1} \text{ Na} + 5.982430704 \cdot 10^{-2} \text{ K} - 1.279464174 \cdot 10^{-2} \text{ Fe} + 6.442408358 \cdot 10^{-1} \text{ Na} + 5.982430704 \cdot 10^{-2} \text{ K} - 1.279464174 \cdot 10^{-2} \text{ Fe} + 6.442408358 \cdot 10^{-1} \text{ Na} + 5.982430704 \cdot 10^{-2} \text{ K} - 1.279464174 \cdot 10^{-2} \text{ Fe} + 6.442408358 \cdot 10^{-1} \text{ Na} + 5.982430704 \cdot 10^{-2} \text{ K} - 1.279464174 \cdot 10^{-2} \text{ Fe} + 1.2794641$
$8.562195211 \cdot 10^{-1}$ Zn + 1.163320738
$Mg = -6.35390668 \cdot 10^{-3} \text{ Ca-} 6.318568338 \cdot 10^{-3} \text{ Fe-} 8.627021799 \cdot 10^{-2} \text{ Na-} 4.557614174 \cdot 10^{-2} \text{ K} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2} \text{ Zn} + 1.396812679 \cdot 10^{-2$
$1.750846518 \cdot 10^{-2} \text{ Cu} + 1.532218107$
$Ca = -2.571958092 \cdot 10^{-1} Fe- 2.711071288 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 2.707204865 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Zn + 9.431309118 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Cu - 7.128587394 Na - 1.705683729 K - 9.431309118 \cdot 10^{-2} Cu - 7.128587394 Na - 1.7056874 K - 9.4578874 K - 9.457887$
Mg + 56.92535167
Fe= -5.638771478 Na- 3.657361698 K- 1.259088853 Zn - 1.549212103·10 ⁻¹ Cu - 28.72630293 Mg - 1.04222693 Ca +
186.9857094
$Na = -0.484812983 K + 7.651382081 \cdot 10^{-2} \text{ Zn} + 9.50297564 \cdot 10^{-2} \text{ Cu} - 4.778048563 Mg - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.338344614 \cdot 10^{-1} \text{ Ca} - 1.33834461$
6.869309898·10 ⁻² Fe+ 16.3834748

Table-5. Regression coefficients in paralytic patients and controls				
S.N.	FORMULA	PARALYSIS	CONTROL	
1	$b_{ZnCu.MgCaFeNaK} = -\frac{\sigma_1}{\sigma_2} \times \frac{\omega_2}{\omega_{11}}$	-10.29	-18.195	
2	$b_{ZnMg.CaFeNaKCu} = -\frac{\sigma_1}{\sigma_3} \times \frac{\omega_3}{\omega_{11}}$	1.506	-2.310	
3	$b_{Zn.CaCuMgFeNaK} = -\frac{\sigma_1}{\sigma_4} \times \frac{\omega_{14}}{\omega_{11}}$	-0.2830	0.1183	
4	$b_{ZnFe.CuMgCaNaK} = -\frac{\sigma_1}{\sigma_5} \times \frac{\omega_{15}}{\omega_{11}}$	1.68	0.5301	
5	$b_{ZnNa.CuMgCaFeK} = -\frac{\sigma_1}{\sigma_6} \times \frac{\omega_{16}}{\omega_{11}}$	-0.00785	-0.2255	
6	$b_{ZnK.CuMgCaFeNa} = -\frac{\sigma_1}{\sigma_7} \times \frac{\omega_{17}}{\omega_{11}}$	-0.330	-1.987	

Table-6. Correlation coefficients in paralytic and controls				
S.N.	Correlation	Value of the	Value of the Correlation	
	Coefficient	Correlation in Paralysis	in Control	
1	r ZnCu	0.2144	0.1896	
2	r ZnMg	-0.7555	-0.3341	
	r ZnCa	-0.4263	-0.0727	
4	r ZnFe	0.0991	0.4814	
5	r _{ZnNa}	0.2909	0.0128	
6	r ZnK	0.1673	-0.0576	
7	r CuZn	0.2144	0.1896	
8	r СиМg	-0.0494	0.5086	
9	r CuCa	-0.1984	0.6224	
10	r СиFe	0.3201	0.2159	
11	r CuNa	-0.0697	0.1926	
12	Г СиК	-0.3838	-0.3532	

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13	r_{MgZn}	-0.7555	-0.3341
14	r MgCu	-0.0494	0.5086
15	r _{MgCa}	0.4637	0.0506
16	r MgFe	-0.1042	0.1536
17	r MgNa	-0.6033	0.1437
18	r _{MgK}	0.0521	0.0433
19	r _{CaZn}	-0.4263	-0.0727
20	r са2н	0.3201	0.6224
21	r CaCu	0.4637	0.0506
22		-0.2614	0.3362
23		-0.4252	0.2080
	r CaNa	-0.3434	-0.2873
25	r n z	0.0991	-0.4814
26+	r FeZn r FeCu	0.3201	0.2159
27	r FeMg	-0.1042	0.1536
28	r FeCa	-0.2614	0.3362
29	r FeNa	-0.0718	-0.3674
30	r _{FeK}	-0.3437	-0.3441
31	r NaZn	0.2909	0.0128
32	r _{NaCu}	-0.0697	0.1926
33	r NaMg	-0.6033	0.1437
34	r _{NaCa}	-0.4252	0.208
35	r _{NaFe}	-0.0718	-0.3674
36	r _{NaK}	-0.1955	0.2935
37	r _{KZn}	0.1673	-0.0576
38	r _{KCu}	-0.3838	-0.3532
39	r кмg	0.0521	0.0433
40	r KCa	-0.3434	-0.2873
41	r KFe	-0.3437	-0.3441
42	r KNa	-0.1955	0.2935
43	r ZnZn	1	1
44	Г СиСи	1	1
45	r _{MgMg}	1	1
46	r CaCa	1	1
47	r FeFe	1	1
48	r NaNa	1	1
49	r кк	1	1

5. Discussion and Conclusion

All the trace elements were higher in paralytic samples compared to normal healthy controls .On the basis of statistical analysis we have measured regression and correlation coefficients between different trace elements like Na, K, Ca, Mg, Zn, Cu and Fe in normal samples .Fluctuations in correlations between two elements were also seen .It is suggested that more precisely observations are required for further investigations .We have tried to give a brief idea of regression equations between all 7 elements and a good support is in the direction of implementing these data in manufacturing medicines in the present work.

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References

- [1] Personalhealthfacts.com, 1985. *The importance of trace minerals*. NFM'S, Nutrition Science, News, pp. 1-42.
- [2] Zumkley, H., 1987. "Trace elements in medicine." *Fresenius z. Anal. Chem.*, vol. 327, p. 6.
- [3] Davies, I. J. J., 1972. *Clinicalsignificance of the essential biological metals*. London: William Heineman. p. 48.
- [4] Prasad, A. S. and Oberleas, D., 1970. "Binding of zinc to amino acids and serum proteins in vitro." *J. Lab. Clin. Med.*, vol. 76, pp. 416-452.
- [5] McCance and Widdowson, E. M., 1937. "Absorption and excretion of iron." *Lancet*, vol. 2, pp. 680-684.
- [6] Heaney, R. P., Saville, P. D., and Recker, R. R., 1975. "Calcium absorption as a function of calcium intake." *J. Lab. Clin. Med.*, vol. 85, pp. 881-890.
- [7] Ireland, P. and Fordtran, J. S., 1973. "Effect of dietary calcium and age on jejunal calcium absorption in humans studied by intestinal perfusion." *J. Clin. Investig.*, vol. 52, pp. 2672-2681.
- [8] Wilkinson, R., 1976. "Absorption of calcium, phosphorus and magnesium calcium phosphate and magnesium metabolism (Nordin, B. E. C. eds.) Edinberg." *Churchill Livingstone*, pp. 36-112.
- [9] Gupta, S. C. and Kapoor, V. K., 2013. *Fundamentals of mathematical statistics, s. Chand and sons*. Educational Publishers New Delhi, pp. 10.2-11.26.